



# An unusual silver–ethynide polymeric chain containing centrosymmetric $\text{Ag}_{14}$ cluster segments stabilized by mixed carboxylate ligands

Yun-Peng Xie <sup>a,b</sup>, Shael A. Al-Thabaiti <sup>c</sup>, Mohamed Mokhtar <sup>c</sup>, Thomas C.W. Mak <sup>a,c,\*</sup>

<sup>a</sup> Department of Chemistry and Center of Novel Functional Molecules, The Chinese University of Hong Kong, Shatin, New Territories, Hong Kong, China

<sup>b</sup> College of Materials Science and Engineering, Huazhong University of Science and Technology, Wuhan, Hubei Province, People's Republic of China

<sup>c</sup> Department of Chemistry, King Abdulaziz University, Jeddah 21589, Saudi Arabia

## ARTICLE INFO

### Article history:

Received 26 January 2013

Accepted 27 February 2013

Available online 5 March 2013

### Keywords:

Metal–ligand supramolecular synthon

Organosilver complex

Polymeric coordination chain

Silver(I) ethynide cluster

## ABSTRACT

From the reaction of  $\text{AgC}\equiv\text{C}^t\text{Bu}$  with 9-hydroxy-9-fluorene-carboxylic acid (HL) and  $\text{AgCF}_3\text{CO}_2$ , we have isolated a new type of organosilver(I) coordination polymeric chain, namely  $\{[\text{Ag}_{14}(\text{C}\equiv\text{C}^t\text{Bu})_8(\text{L})_2(\text{CF}_3\text{CO}_2)_4]\cdot 2\text{CH}_3\text{OH}\}_n$ , which contains centrosymmetric  $[\text{Ag}_{14}(\text{C}\equiv\text{C}^t\text{Bu})_8(\text{CF}_3\text{CO}_2)_2]^{4+}$  cluster segments bridged by pairs of L and trifluoroacetate ligands.

© 2013 Elsevier B.V. All rights reserved.

The chemistry of metal ethynyl complexes is of great interest because of their structural diversity and technological application as precursors of nonlinear optical materials, luminescent probes, and rigid-rod molecular wires [1–4]. Among a wide range of candidate metals, the Group 11 elements (Cu, Ag and Au) are frequently employed in the syntheses of novel ethynyl complexes due to their enhanced metalophilic interaction that overcomes repulsion between cationic metal centers [5–11]. The silver(I) ethynyl derivatives are less explored because of their sensitivity to light and tendency to yield highly insoluble polymers/oligomers. Recently, we and other groups have prepared and structurally characterized a series of silver(I)–ethynide coordination polymeric networks [12–17] and high-nuclearity clusters [18–31] based on the multinuclear metal–ligand supramolecular synthon  $\text{R}-\text{C}\equiv\text{C} \supset \text{Ag}_n$  ( $n = 3, 4, 5$ ; R = alkyl, aryl, heteroaryl).

In 1999 the reaction of  $[\text{AgC}\equiv\text{C}^t\text{Bu}]_n$  with  $\text{AgBF}_4$  in molar ratio 2:1 was reported to yield a polymeric undulated ribbon structure  $[\text{Ag}_3(\text{C}\equiv\text{C}^t\text{Bu})_2^+]_n$  [32]. Envisioning that such soluble species might be good precursors for the preparation of new silver(I)–ethynide complexes, we have developed a synthetic strategy by introducing phosphonate and phosphonate-functionalized oxovanadate or oxovanadate as capping ligands for trapping oligomeric silver(I)–ethynide clusters. Recently we isolated a series of mixed-metal giant clusters with the silver(I)–*tert*-butylethynide moiety and various phosphonate-functionalized oxovanadate building blocks as their surface components, as well as different

numbers and kinds of anionic species as their encapsulated templates: for example  $\{(\text{NO}_3)_2\text{Ag}_{16}(\text{C}\equiv\text{CPh})_4(\text{C}\equiv\text{C}^t\text{BuPO}_3)_4\text{V}_4\text{O}_8\}_2(\text{DMF})_6(\text{NO}_3)_2$  [31] and  $\{\text{Cl}_2\text{Ag}_{21}(\text{C}\equiv\text{C}^t\text{Bu})_9\{(\text{C}\equiv\text{C}^t\text{BuPO}_3)_3\text{V}_3\text{O}_6(\text{OH})\}_2\{(\text{C}\equiv\text{C}^t\text{BuPO}_3)\text{VO}_2(\text{OH})\}(\text{MeOH})_2(\text{H}_2\text{O})_2\}$  [30]. We also employed neat *tert*-butylphosphonic acid as a precursor and successfully generated new silver(I) ethynide networks and clusters, for example  $3\text{AgC}\equiv\text{CPh}\cdot\text{Ag}_2^t\text{BuPO}_3\cdot\text{Ag}^t\text{BuPO}_3\text{H}\cdot 2\text{AgNO}_3$  and  $\{[\text{Ag}_8(\text{Cl@Ag}_{14})](\text{C}\equiv\text{C}^t\text{Bu})_{14}(\text{C}\equiv\text{C}^t\text{BuPO}_3)_2\text{F}_2(\text{H}_2\text{O})_2\}\text{BF}_4\cdot 3.5\text{H}_2\text{O}$  [16].

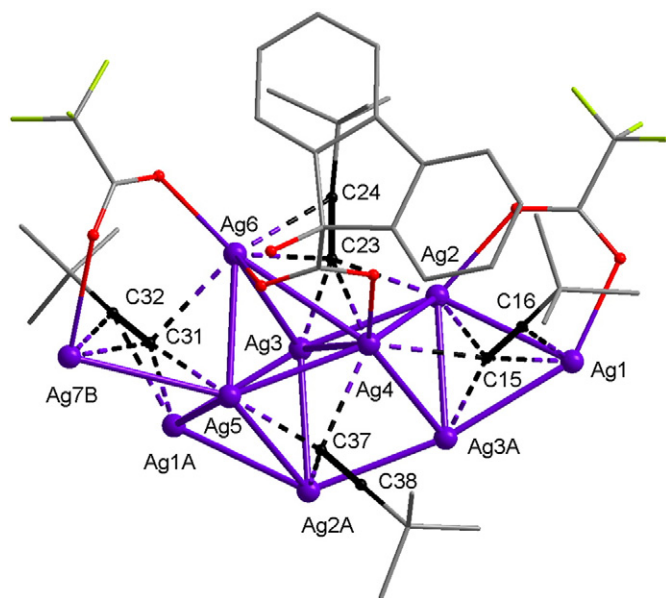
In the present work, we report our successful combination of 9-hydroxy-9-fluorene-carboxylic acid (HL) with the supramolecular synthon  ${}^t\text{BuC}\equiv\text{C} \supset \text{Ag}_n$  ( $n = 3, 4$ ) to generate a novel organosilver coordination polymer, namely  $\{[\text{Ag}_{14}(\text{C}\equiv\text{C}^t\text{Bu})_8(\text{L})_2(\text{CF}_3\text{CO}_2)_4]\cdot 2\text{CH}_3\text{OH}\}_n$  (**1**) [33], which has been characterized by elemental analysis, infrared spectroscopy and single-crystal X-ray analysis [34]. HL is known to function as a plant growth regulator, which restrains the generation of abscisic acid and regulate plant root calcium transport [35]. Furthermore, in the anionic L ligand, the hydroxy and carboxylate groups provide different potential coordination sites for metal coordination.

The reaction of HL with  $\text{AgC}\equiv\text{C}^t\text{Bu}$  and  $\text{AgCF}_3\text{CO}_2$  in  $\text{CH}_3\text{OH}$  yielded colorless prismatic crystals of **1**. An IR vibration band at  $2005\text{ cm}^{-1}$  confirmed the presence of the  $\text{C}\equiv\text{C}$  functional group. Of the four independent *tert*-butylethynide ligands in the asymmetric unit, the ethynide group composed of C37 and C38 is bound to a silver triangle in the  $\mu_3-\eta^1, \eta^1, \eta^1$  coordination mode. The remaining three ethynide groups (C15=C16, C23=C24 and C31=C32) are each capped by a butterfly-shaped  $\text{Ag}_4$  basket in the  $\mu_4-\eta^1, \eta^1, \eta^1, \eta^2$  or  $\mu_4-\eta^1, \eta^1, \eta^2, \eta^2$  coordination mode, as shown in Fig. 1.

Two  $\text{Ag}_3$  triangles and six  $\text{Ag}_4$  baskets coalesce by sharing vertices to form a centrosymmetric  $\text{Ag}_{14}({}^t\text{BuC}\equiv\text{C})_8$  cluster, which is further stabilized by a pair of  $\text{CF}_3\text{CO}_2$  ligands with  $\text{Ag}-\text{O}$  bond distances of

\* Corresponding author at: Department of Chemistry and Center of Novel Functional Molecules, The Chinese University of Hong Kong, Shatin, New Territories, Hong Kong, China. Tel.: +852 3943 6279; fax: +852 2603 5057.

E-mail address: [tcwmak@cuhk.edu.hk](mailto:tcwmak@cuhk.edu.hk) (T.C.W. Mak).



**Fig. 1.** Coordination environment of silver(I) atoms surrounded by four independent *tert*-butylethyne, one 9-hydroxy-9-fluorene-carboxylate, and two trifluoroacetate ligands in  $[\{Ag_{14}(C\equiv C^tBu)_8(L)_2(CF_3CO_2)_4\} \cdot 2CH_3OH]_n$  (**1**). The  $C\equiv C$  triple bond is represented by a bold rod. The  $Ag-C$  bonds are indicated by broken lines. All hydrogen atoms and methanol molecules have been omitted for clarity. Color codes: Ag, violet; C, gray; O, red; and F, lime.  $Ag1A-C31 = 2.325(8)$ ,  $Ag1-C15 = 2.334(9)$ ,  $Ag1-C16 = 2.647(9)$ ,  $Ag1A-C32 = 2.679(7)$ ,  $Ag2A-C37 = 2.349(9)$ ,  $Ag2-C23 = 2.521(9)$ ,  $Ag3-C23 = 2.134(7)$ ,  $Ag3A-C15 = 2.151(8)$ ,  $Ag4-C37 = 2.429(7)$ ,  $Ag4-C15 = 2.438(9)$ ,  $Ag4-C23 = 2.519(6)$ ,  $Ag5-C37 = 2.150(8)$ ,  $Ag5-C31 = 2.195(7)$ ,  $Ag6-C23 = 2.376(8)$ ,  $Ag6-C31 = 2.494(7)$ ,  $Ag6-C24 = 2.545(8)$ ,  $Ag7B-C31 = 2.256(8)$ ,  $Ag7B-C32 = 2.665(7)$ . Symmetry codes: A  $1-x, -y, -z$ ; and B  $2-x, -y, -z$ .

2.317(5) and 2.369(6) Å (Fig. 2). The resulting  $Ag_{14}(^tBuC\equiv C)_8(CF_3CO_2)_2$  cluster segments are bridged by a series of L and additional  $CF_3CO_2$  ligands to engender an infinite coordination chain in the direction of the *a* axis (Fig. 3). The L ligand has its carboxylate terminal bridging two adjacent silver atoms, and the 9-hydroxy group functions as a monodentate ligand to coordinate one silver atom, with  $Ag-O$  distances

varying from 2.343(6) to 2.589(5) Å. However, the bridging  $CF_3CO_2$  ligand coordinates to four silver atoms with  $Ag-O$  bond lengths ranging from 2.335(6) to 2.715(6) Å.

The argentophilic  $Ag-Ag$  bond distances lie in the range 2.899(2)–3.325(2) Å. Additionally, the crystal structure contains two methanol solvate molecules in the unit cell. Fig. 4 shows the crystal structure with only van der Waals interaction between the coordination polymeric chains.

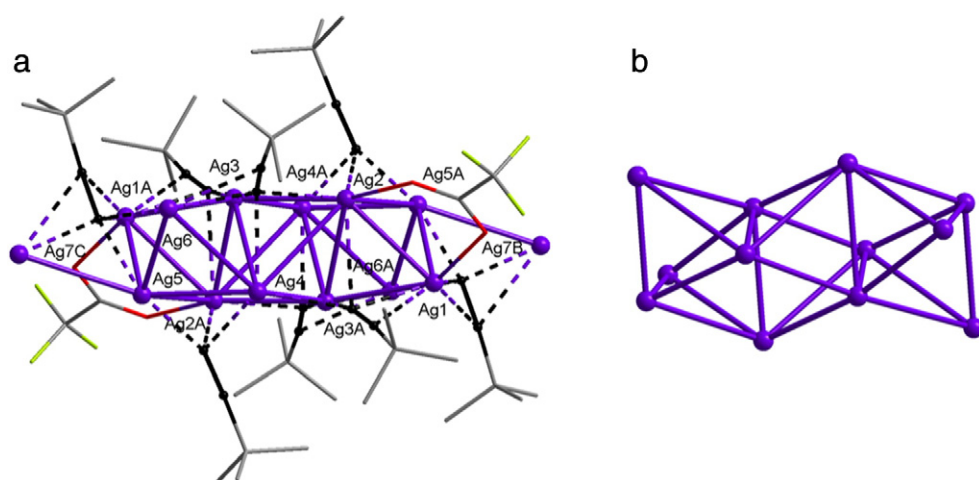
Notably, the bar-like  $Ag_{14}$  cluster in **1** differs markedly from previously reported rhombic-dodecahedral silver(I)-ethynide cages, e.g.  $[Cl@Ag_{14}(C\equiv C^tBu)_{12}]OH$  [19] and  $[Cl@Ag_{14}(C\equiv C_{cyclohexyl})_{12}]Cl$  [28] with encapsulated chloride ions, which have been employed as precursors to synthesize the corresponding iso-structural “cluster-within-cluster” compounds  $[Cl_6Ag_8@Ag_{30}(^tBuC\equiv C)_{20}(ClO_4)_{12}] \cdot Et_2O$  and  $[Cl_6Ag_8@Ag_{30}(chxC\equiv C)_{20}(ClO_4)_{10}](ClO_4)_2 \cdot 1.5Et_2O$  that bear the same pseudo- $O_h$  cationic  $Cl_6Ag_8$  inner core [28].

The anionic L ligand, with its bulky hydrophobic, outward-extending fluorenyl group, connects adjacent high-nuclearity silver(I)-ethynide cluster segments to generate the coordination polymer chain in **1**. In contrast, silver trifluoroacetate plays a crucial dual role in the present synthesis: (i) as a bridging ligand that stabilizes each  $Ag_{14}$  cluster segment, and (ii) cooperates with the L ligand to link up adjacent  $Ag_{14}$  cluster segments. Notably, thus far our attempts to prepare related silver(I)-ethynide complexes using other soluble silver salts such as nitrate, tetrafluoroborate and triflate were unsuccessful, which always yielded an intractable pale solid residue.

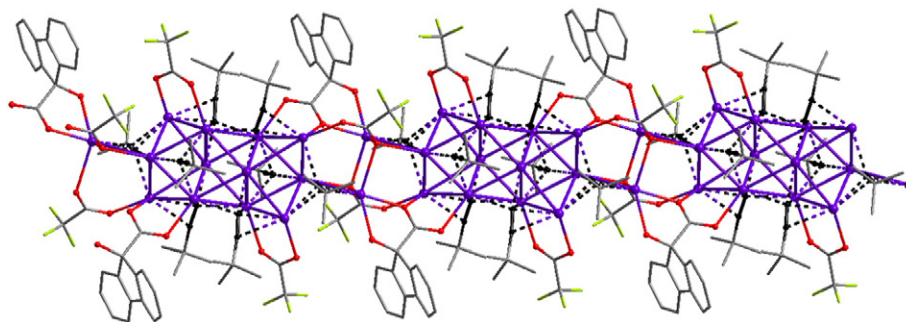
Combining the metal-ligand supramolecular synthon  $^tBuC\equiv C \supset Ag_n$  ( $n = 3, 4$ ) (generated *in situ* from a solution of  $^tBuC\equiv CAg$  and  $AgCF_3CO_2$  in methanol) and 9-hydroxy-9-fluorene-carboxylic acid, we have isolated and structurally characterized a new kind of silver(I)-ethynide coordination polymeric complex, in which the 9-hydroxy-9-fluorene-carboxylate and trifluoroacetate ligands bridge two novel condensed  $Ag_{14}$  clusters to form a coordination chain architecture. The present study offers broad prospects to future synthetic exploration of silver-ethynide complexes bearing different varieties of carboxylates as structural components.

#### Acknowledgments

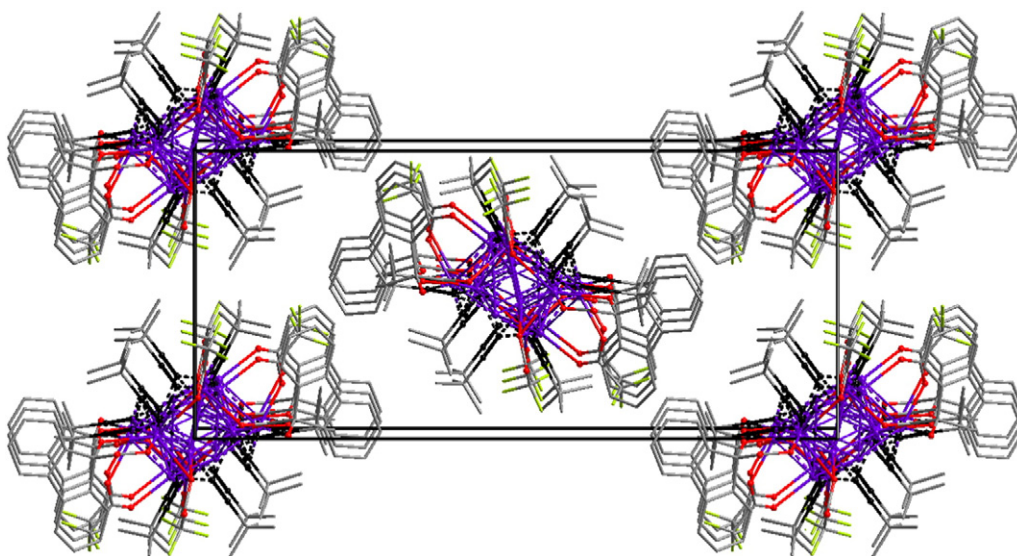
This work was supported by the Wei Lun Foundation, grant no. 27-3-1432/HiCi of King Abdulaziz University, and a Postdoctoral



**Fig. 2.** (a) The centrosymmetric  $[Ag_{14}(C\equiv C^tBu)_8(CF_3CO_2)_2]^{4+}$  cluster skeleton in complex **1**. H atoms are omitted for clarity. The carbon atoms of the ethynide group are represented as small black balls, and their bonds to silver atoms are indicated by broken lines. Atoms  $Ag7B$  and  $Ag7C$  are symmetry equivalents of  $Ag7$ . (b) Perspective view of the center  $Ag_{12}$  portion in  $[Ag_{14}(C\equiv C^tBu)_8(CF_3CO_2)_2]^{4+}$  cluster skeleton. Color codes: Ag, violet; C, gray; O, red; and F, lime.  $Ag1-Ag5A = 2.899(2)$ ,  $Ag1-Ag3A = 2.985(2)$ ,  $Ag1-Ag2 = 3.076(2)$ ,  $Ag2-Ag3 = 2.945(2)$ ,  $Ag2-Ag3A = 3.064(3)$ ,  $Ag2-Ag5A = 3.115(3)$ ,  $Ag2-Ag4 = 3.317(2)$ ,  $Ag3-Ag4 = 2.983(2)$ ,  $Ag3-Ag4A = 2.995(2)$ ,  $Ag3-Ag6 = 3.278(3)$ ,  $Ag4-Ag5 = 3.000(2)$ ,  $Ag4-Ag6 = 3.231(2)$ ,  $Ag5-Ag6 = 2.938(3)$ ,  $Ag5-Ag7B = 3.325(2)$ . Symmetry codes: A  $1-x, -y, -z$ ; B  $1+x, y, z$ ; and C  $2-x, -y, -z$ .



**Fig. 3.** Coordination polymeric chain in complex **1**. Each C=C triple bond is represented by a bold rod. The Ag–C bonds are indicated by broken lines. All hydrogen atoms and methanol molecules have been omitted for clarity. Color codes: Ag, violet; C, gray; O, red; and F, lime.



**Fig. 4.** Packing of coordination polymeric chains in the crystal structure of **1** viewed parallel to the *a* axis. All methanol molecules have been omitted for clarity. Color codes: Ag, violet; C, gray; O, red; and F, lime.

Research Fellowship award to Y.-P. Xie by The Chinese University of Hong Kong.

#### Appendix A. Supplementary material

CCDC 907576 contains the supplementary crystallographic data for this paper, which can be obtained free of charge from the Cambridge Crystallographic Data Centre via [www.ccdc.cam.ac.uk/data\\_request/cif](http://www.ccdc.cam.ac.uk/data_request/cif). Supplementary data associated with this article can be found, in the online version, at <http://dx.doi.org/10.1016/j.inoche.2013.02.019>.

#### References

- [1] M.I. Bruce, Organometallic chemistry of vinylidene and related unsaturated carbenes, *Chem. Rev.* 91 (1991) 197–257.
- [2] F. Paul, C. Lapinte, Organometallic molecular wires and other nanoscale-sized devices: an approach using the organoiron (dppe)Cp\*Fe building block, *Coord. Chem. Rev.* 178–180 (1998) 431–509.
- [3] V.W.-W. Yam, Molecular design of transition metal alkynyl complexes as building blocks for luminescent metal-based materials: structural and photophysical aspects, *Acc. Chem. Res.* 35 (2002) 555–563.
- [4] V.W.-W. Yam, E.C.-C. Cheng, Highlights on the recent advances in gold chemistry—a photophysical perspective, *Chem. Soc. Rev.* 37 (2008) 1806–1813.
- [5] M. Jansen, Homoatomic d<sup>10</sup>–d<sup>10</sup> interactions: their effects on structure and chemical and physical properties, *Angew. Chem. Int. Ed.* 26 (1987) 1098–1110.
- [6] P. Pykkö, Strong closed-shell interactions in inorganic chemistry, *Chem. Rev.* 97 (1997) 597–636.
- [7] P. Pykkö, Theoretical chemistry of gold, *Angew. Chem. Int. Ed.* 43 (2004) 4412–4456.
- [8] V.W.-W. Yam, K.M.-C. Wong, Luminescent molecular rods-transition-metal alkynyl complexes, *Top. Curr. Chem.* 257 (2005) 1–32.
- [9] R.J. Puddephatt, Macrocycles, catenanes, oligomers and polymers in gold chemistry, *Chem. Soc. Rev.* 37 (2008) 2012–2027.
- [10] H.L. Hermann, G. Boche, P. Schwerdtfeger, Metallophilic interactions in closed-shell copper(I) compounds—a theoretical study, *Chem. Eur. J.* 7 (2001) 5333–5342.
- [11] C.J. Adams, M.I. Bruce, E. Horn, B.W. Skelton, E.R.T. Tiekink, A.H. White, Cluster chemistry. Part 91. Clusters derived from 1,4-bis (diphenylphosphino)buta-1,3-diyne and their pyrolysis products: a route to complexes containing the tetracarbon ligand, *J. Chem. Soc. Dalton Trans.* (1993) 3299–3312.
- [12] L. Zhao, W.-Y. Wong, T.C.W. Mak, Novel  $\mu_6$ -coordination modes of aryl and alkyl ethynides and classification of metal–ligand interactions in silver(I) complexes, *Chem. Eur. J.* 12 (2006) 4865–4872.
- [13] S.-Q. Zang, J. Han, T.C.W. Mak, Silver(I)–organic networks assembled with the flexible prop-2-ynoxybenzene ligand: *in situ* recrystallization and unusual silver–aromatic interaction, *Organometallics* 28 (2009) 2677–2683.
- [14] B. Li, S.-Q. Zang, R. Liang, Y.-J. Wu, T.C.W. Mak, Silver(I)–organic networks assembled with propargyl-functionalized di- and trihydroxybenzenes, *Organometallics* 30 (2011) 1710–1718.
- [15] Y. Zhao, P. Zhang, B. Li, X.G. Meng, T.L. Zhang, Two- and three-dimensional silver(I)–organic networks generated from mono- and dicarboxylphenylethyne, *Inorg. Chem.* 50 (2011) 9097–9105.
- [16] Y.-P. Xie, T.C.W. Mak, Silver(I) ethynide coordination networks and clusters assembled with *tert*-butylphosphonic acid, *Inorg. Chem.* 51 (2012) 8640–8642.
- [17] P.-S. Cheng, S. Marivel, S.-Q. Zang, G.-G. Gao, T.C.W. Mak, Argentophilic infinite chain, column, and layer structures assembled with the multinuclear silver(I)–phenylethyne supramolecular synthon, *Cryst. Growth Des.* 12 (2012) 4519–4529.
- [18] D. Rais, J. Yau, D.M.P. Mingos, R. Vilar, A.J.P. White, D.J. Williams, Anion-templated syntheses of rhombohedral silver-alkynyl cage compounds, *Angew. Chem. Int. Ed.* 40 (2001) 3464–3467.
- [19] D. Rais, D.M.P. Mingos, R. Vilar, A.J.P. White, D.J. Williams, Directing role of anions in the syntheses of the silver-alkynyl cages [Ag<sub>14</sub>(C≡C<sup>t</sup>Bu)<sub>12</sub>X][BF<sub>4</sub>] (X = F, Cl, Br)

- and silver-alkynyl polymers  $[Ag_3(C\equiv CtBu)_2(X)]_n$  ( $X = Tos, NO_3$ ), *J. Organomet. Chem.* 652 (2002) 87–93.
- [20] S.-D. Bian, H.-B. Wu, Q.-M. Wang, A facile template approach to high-nuclearity silver(I) alkynyl clusters, *Angew. Chem. Int. Ed.* 48 (2009) 5363–5365.
- [21] S.-D. Bian, J.-H. Jia, Q.-M. Wang, High-nuclearity silver clusters templated by carbonates generated from atmospheric carbon dioxide fixation, *J. Am. Chem. Soc.* 131 (2009) 3422–3423.
- [22] J. Qiao, K. Shi, Q.-M. Wang, A giant silver alkynyl cage with sixty silver(I) ions clustered around polyoxometalate templates, *Angew. Chem. Int. Ed.* 49 (2010) 1765–1767.
- [23] P. Putaj, F. Lefebvre, Polyoxometalates containing late transition and noble metal atoms, *Coord. Chem. Rev.* 255 (2011) 1642–1685.
- [24] C.-Y. Gao, L. Zhao, M.-X. Wang, Designed synthesis of metal cluster-centered pseudo-rotaxane supramolecular architectures, *J. Am. Chem. Soc.* 133 (2011) 8448–8451.
- [25] F. Gruber, M. Jansen,  $\{[Ag_{42}(CO_3)(C\equiv C^tBu)_{27}(CH_3CN)_2][CoW_{12}O_{40}]_2\}[BF_4]$ : an intercluster sandwich compound, *Angew. Chem. Int. Ed.* 49 (2010) 4924–4926.
- [26] G.-G. Gao, P.-S. Cheng, T.C.W. Mak, Acid-induced surface functionalization of polyoxometalate by enclosure in a polyhedral silver-alkynyl cage, *J. Am. Chem. Soc.* 131 (2009) 18257–18259.
- [27] Y.-P. Xie, T.C.W. Mak, Silver(I)-ethynide clusters constructed with phosphonate-functionized polyoxovanadates, *J. Am. Chem. Soc.* 133 (2011) 3760–3763.
- [28] S.C.K. Hau, P.-S. Cheng, T.C.W. Mak, Enlargement of globular silver alkynide cluster via core transformation, *J. Am. Chem. Soc.* 134 (2012) 2922–2925.
- [29] Y.-P. Xie, T.C.W. Mak, A pyrovanadate-templated silver(I)-ethynide cluster circumscribed by macrocyclic polyoxovanadate(V), *Chem. Commun.* (2012) 1123–1125.
- [30] Y.-P. Xie, T.C.W. Mak, High-nuclearity silver ethynide clusters assembled with phosphonate and metavanadate precursors, *Angew. Chem. Int. Ed.* 51 (2012) 8783–8786.
- [31] Y.-P. Xie, T.C.W. Mak, Silver-ethynide clusters with oxovanadate components, *J. Clust. Sci.* 23 (2012) 727–736.
- [32] K.A. Al-Farhan, M.H. Ja'far, O.M. Abu-Salah, The synthesis, identification, and X-ray structure of a novel cationic alkynyl silver cluster polymer  $[Ag_5(^tBuC\equiv C)_2^+]_n$ , *J. Organomet. Chem.* 579 (1999) 59–62.
- [33] HL (0.023 g, 0.010 mmol) was dissolved in  $CH_3OH$  (4 mL). Then  $^tBuC\equiv CAg$  (0.095 g, 0.500 mmol) and  $AgCF_3CO_2$  (0.022 g, 0.100 mmol) were added under stirring. After 30 min, a clear solution was collected by filtration. Slow evaporation of the clear solution afforded colorless prismatic crystals **1** in ca. 16%. Elemental analysis (%) calcd for  $C_{43}H_{49}Ag_7F_6O_8$ : C 33.04, H 3.16; found: C 33.35, H 3.31; selected IR data (KBr): 2005 ( $C\equiv C$ ).
- [34] Crystallographic data. Complex **1**: monoclinic,  $a = 13.625(1)$ ,  $b = 29.180(3)$ ,  $c = 13.767(1)$  Å,  $\beta = 111.262(1)$  °,  $V = 5101(8)$  Å<sup>3</sup>,  $T = 293$  K, space group  $P2_1/c$ ,  $Z = 4$ ,  $\lambda = 0.71073$  Å,  $\rho = 2.034$  cm<sup>-3</sup>,  $\mu(MoK\alpha) = 2.699$  mm<sup>-1</sup>,  $R_1 = 0.0462$ ,  $wR_2 = 0.1366$  for  $I > 2\sigma(I)$ , GOF = 1.117. Crystal data were collected on a Bruker Smart Apex II CCD diffractometer with Mo-K $\alpha$  radiation ( $\lambda = 0.71073$  Å). The structures were solved by direct methods with SHELXS-97 and refined by full-matrix least-squares techniques using the SHELXL-97 program [36] within WINGX [37].
- [35] F. Migliaccio, A.W. Galston, On the nature and origin of the calcium asymmetry arising during gravitropic response in etiolated pea epicotyls, *Plant Physiol.* 85 (1987) 542–547.
- [36] G.M. Sheldrick, SHELXL-97, Program for Crystal Structure Solution, University of Göttingen: Göttingen, Germany, 1997.
- [37] L.J. Farrugia, WINGX, A Windows Program for Crystal Structure Analysis, University of Glasgow, UK, 1988.